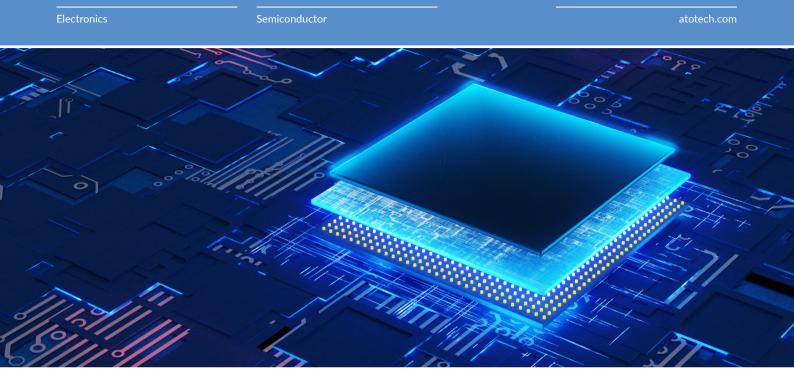
Spherolyte[®] Cu DB Next generation Cu-to-Cu direct bonding





Low temperature annealing Cu for next generation bonding

Copper direct bonding for advanced packaging

Cu-to-Cu direct bonding promises to be one of the key technologies for the manufacturing of next generation semiconductor devices. It allows enhanced device performance due to superior electrical performance and simultaneous smaller form factors and higher reliabilities. Additionally, wafer-to-wafer bonding enables cost advantages and improved production efficiencies.

Spherolyte[®] Cu DB meets the requirements of next generation Cu-to-Cu bonding, for which relatively low bonding temperatures ≤200 °C are required.

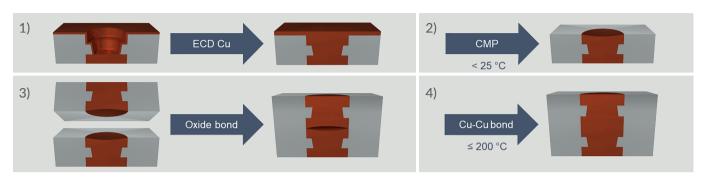
Features and benefits

- Bond formations at low temperatures (≤ 200 °C)
- Deposition of pure copper
- Superior via filling capabilities
- High uniformity
- Good ductility
- Good electromigration performance



Next generation Cu-to-Cu direct bonding

Cu-to-Cu direct bonding sequence based on metastable Cu



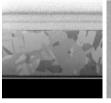
1) Filling of vias with electrolytic metastable Cu; 2) chemical mechanical polishing to remove Cu cover layer; 3) Preliminary dielectrics bonding; 4) annealing step at temperatures ≤ 200 °C to allow Cu-to-Cu bond formation over interface.

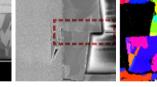
Bondformation by metastable copper

Spherolyte® Cu DB is our new process for the deposition of metastable Cu, which allows to execute Cu-to-Cu direct bonding at low temperatures.

Standard processes deposit coarse grained Cu which under low-temperatures does not change its structures and hence requires higher bonding temperatures. Spherolyte[®] Cu DB deposits pure, metastable Cu which recrystallizes at temperatures ≤ 200 °C, leading to a grain growth of the Cu interface and hence a firm bonding step.

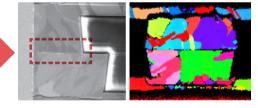
Non-optimized process (stable, coarse-grained microstructure)





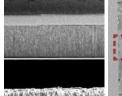
Before bonding

After dielectrics bonding



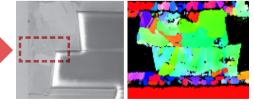
Simulated package formation

Optimized process (metastable microstructure)



Before bonding

After dielectrics bonding



Simulated package formation



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